

C(11)	0.4778 (3)	0.3602 (5)	0.1611 (2)	5.8 (2)
C(12)	0.0872 (4)	0.2203 (5)	-0.0201 (2)	6.4 (2)
C(13)	0.3970 (4)	0.2872 (5)	-0.0376 (3)	6.2 (2)
C(14)	0.3508 (6)	0.0245 (5)	0.0819 (3)	8.0 (3)

Table 2. Selected geometric parameters (\AA , $^\circ$)

Al—C(14)	1.956 (5)	Al—C(13)	1.962 (4)
Al—C(12)	1.971 (4)	Al—N(1)	2.050 (2)
N(1)—C(4)	1.488 (3)	N(1)—C(1)	1.494 (3)
N(1)—C(11)	1.494 (4)	N(2)—C(5)	1.405 (3)
N(2)—C(3)	1.461 (4)		
C(14)—Al—C(13)	115.1 (2)	C(14)—Al—C(12)	113.6 (2)
C(14)—Al—N(1)	103.6 (2)	C(13)—Al—C(12)	113.9 (2)
C(13)—Al—N(1)	103.6 (1)	C(12)—Al—N(1)	105.4 (1)
C(4)—N(1)—C(1)	106.5 (2)	C(4)—N(1)—C(11)	110.5 (3)
C(4)—N(1)—Al	110.3 (2)	C(1)—N(1)—C(11)	110.0 (2)
C(1)—N(1)—Al	110.2 (2)	C(11)—N(1)—Al	108.3 (2)

The single crystal was sealed in a glass capillary under nitrogen. *CAD-4 Software* (Enraf-Nonius, 1989) was used for data collection and cell refinement. The structure was solved by direct methods using *MITHRIL* (Gilmore, 1983) and subsequent Fourier techniques, and refined by full-matrix least-squares methods. The H atoms were found by difference Fourier methods. H-atom coordinates were used for correction of the structure. All calculations were performed using *TEXSAN* (Molecular Structure Corporation, 1989). *ORTEPII* (Johnson, 1976) was used to produce the figure.

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates, complete geometry and least-squares-planes data have been deposited with the IUCr (Reference: KH1055). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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Trifluoromethyl(triphenylphosphine)gold(I)

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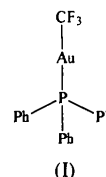
(Received 9 October 1995; accepted 30 October 1995)

Abstract

The title compound, $[\text{Au}(\text{CF}_3)\{\text{P}(\text{C}_6\text{H}_5)_3\}]$, is the first crystallographically characterized mononuclear gold(I) complex with an Au—CF₃ bond. The Au coordination is nearly linear with P—Au—C 178.56(11) $^\circ$, Au—P 2.2859(9) and Au—C 2.045(4) \AA .

Comment

The structures of linear gold(I) complexes Ph_3PAuX have attracted much interest with regard to the type of ligand *X* and the *trans*-positioned Au—P bond (Ahrlund, Dreisch, Norén & Oskarsson, 1987; Barron, Engelhardt, Healy, Oddy & White, 1987; Jones & Lautner, 1988). The complexes with *X* = Cl, Br, I, CN, SCN and OCOCH_3 are effectively isostructural with orthorhombic packing, $P2_12_12_1$, whereas the monoclinic complexes with *X* = ONO_2 and CH_3 differ in their packing. We have investigated the related compound, (I), with *X* = trifluoromethyl, which, according to a search in the Cambridge Structural Database (Allen *et al.*, 1991), is also the first crystallographically characterized example of a mononuclear complex with an Au^{I} —CF₃ bond.



The title molecule shows the expected linear coordination of the Au^{I} atom with a P—Au—C angle of 178.56(11) $^\circ$. The most interesting geometric features are the Au—P and Au—C bond distances. The former, 2.2859(9) \AA , has the same length as those in the complexes with *X* = CN [2.278(2) \AA ; Jones & Lautner, 1988] and *X* = CH_3 [2.279(8) \AA ; Gavens, Guy, Mays & Sheldrick, 1977], although these three ligands differ appreciably in their electronic properties. It has already been stated by Barron *et al.* (1987) that de-

spite the different *trans* influence of X, these groups have rather small effects on Au—P distances in the solid state, but show significant effects on the ³¹P chemical-shift data. Thus, the ³¹P NMR signal for the title compound is $\delta = 38.9$ p.p.m. and for the methyl compound $\delta = 48.0$ p.p.m. (Bruker AMX 300, standard 85% H₃PO₄). The Au—C distance of 2.045 (4) Å is slightly longer than the corresponding bond in the above-mentioned CN complex [2.003 (7) Å]. Comparison with the Au—C bond distance of the CH₃ compound suffers from the high standard deviation of the reported value, 2.124 (28) Å. The only other known structure with an Au—CF₃ bond is an Au^{II} dimer complex and so the considerably longer reported distance of 2.140 (7) Å (Murray, Fackler, Porter, Briggs, Guerra & Lagow, 1987) is not directly comparable. There are no Au...Au interactions, the shortest Au...Au distance being 5.498 (1) Å, nor have any other unusual intermolecular contacts been found.

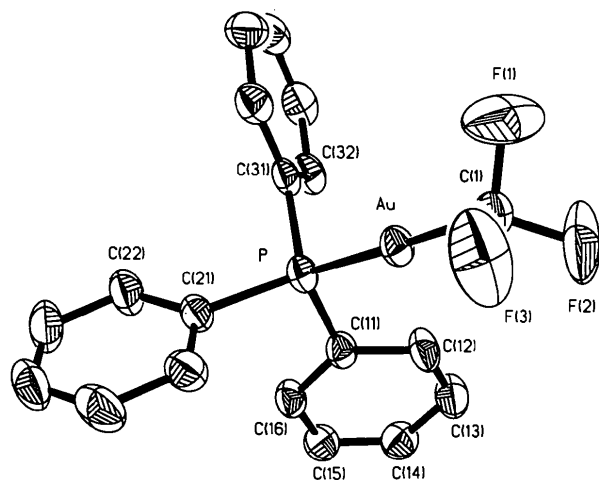


Fig. 1. Molecular structure of the title compound with H atoms omitted. Displacement ellipsoids are plotted at the 50% probability level.

Experimental

The compound was isolated as a by-product from the reaction of triphenylphosphine with Re₂(HgCF₃)(AuPPh₃)(μ-OC₄H₉) in thf solution at 296 K (Siefert, 1994).

Crystal data

[Au(CF₃)(C₁₈H₁₅P)]

$M_r = 528.25$

Monoclinic

$P2_1/c$

$a = 9.1029$ (14) Å

$b = 11.395$ (2) Å

$c = 17.490$ (2) Å

$\beta = 104.953$ (10)°

Mo $K\alpha$ radiation

$\lambda = 0.71073$ Å

Cell parameters from 59

reflections

$\theta = 5$ – 12.5°

$\mu = 8.511$ mm⁻¹

$T = 173$ (2) K

Prism

$V = 1752.8$ (4) Å³

$Z = 4$

$D_x = 2.002$ Mg m⁻³

D_m not measured

Data collection

Siemens P4 diffractometer

ω scans

Absorption correction:

ψ scans (North, Phillips
& Matthews, 1968)

$T_{\min} = 0.682$, $T_{\max} =$
0.937

8307 measured reflections

4013 independent reflections

3266 observed reflections

$[I > 2\sigma(I)]$

Refinement

Refinement on F^2

$R[F^2 > 2\sigma(F^2)] = 0.0227$

$wR(F^2) = 0.0514$

$S = 0.924$

4013 reflections

218 parameters

H atoms riding

$w = 1/[\sigma^2(F_o^2) + (0.0289P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.885$ e Å⁻³

$\Delta\rho_{\min} = -0.757$ e Å⁻³

$0.52 \times 0.26 \times 0.25$ mm

Pale amber

$R_{\text{int}} = 0.0265$

$\theta_{\max} = 27.49^\circ$

$h = -10 \rightarrow 11$

$k = -14 \rightarrow 14$

$l = -21 \rightarrow 22$

3 standard reflections

monitored every 247

reflections

intensity decay: none

Extinction correction:

SHELXL93 (Sheldrick,
1993)

Extinction coefficient:

0.00113 (11)

Atomic scattering factors

from *International Tables*

for *Crystallography* (1992,

Vol. C, Tables 4.2.6.8 and

6.1.1.4)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å²)

$$U_{\text{eq}} = (1/3)\sum_i\sum_j U_{ij}a_i^*a_j^*a_i\cdot a_j$$

	<i>x</i>	<i>y</i>	<i>z</i>	U_{eq}
Au	0.67740 (2)	0.438407 (12)	0.400805 (8)	0.03033 (6)
P	0.58455 (11)	0.26670 (8)	0.43719 (6)	0.0300 (2)
C1	0.7643 (5)	0.5897 (3)	0.3672 (2)	0.0346 (8)
F1	0.8052 (6)	0.5860 (3)	0.3002 (2)	0.116 (2)
F2	0.8875 (4)	0.6311 (3)	0.4190 (2)	0.0891 (11)
F3	0.6733 (4)	0.6819 (3)	0.3565 (3)	0.0987 (12)
C11	0.6521 (4)	0.2336 (3)	0.5414 (2)	0.0317 (8)
C12	0.8012 (4)	0.2638 (4)	0.5800 (2)	0.0419 (10)
C13	0.8636 (5)	0.2298 (4)	0.6577 (2)	0.0477 (10)
C14	0.7774 (5)	0.1645 (3)	0.6970 (2)	0.0417 (9)
C15	0.6287 (5)	0.1359 (3)	0.6596 (2)	0.0402 (9)
C16	0.5659 (4)	0.1702 (3)	0.5821 (2)	0.0367 (8)
C21	0.3790 (4)	0.2620 (3)	0.4137 (2)	0.0322 (8)
C22	0.2986 (4)	0.1582 (4)	0.3938 (3)	0.0449 (10)
C23	0.1422 (5)	0.1571 (5)	0.3795 (3)	0.0547 (12)
C24	0.0661 (5)	0.2587 (5)	0.3856 (3)	0.0566 (12)
C25	0.1436 (5)	0.3627 (5)	0.4042 (3)	0.0570 (12)
C26	0.3012 (5)	0.3656 (4)	0.4177 (2)	0.0433 (10)
C31	0.6453 (4)	0.1431 (3)	0.3872 (2)	0.0324 (8)
C32	0.7366 (5)	0.0530 (3)	0.4278 (3)	0.0397 (9)
C33	0.7860 (5)	-0.0350 (4)	0.3857 (3)	0.0462 (10)
C34	0.7474 (5)	-0.0328 (4)	0.3038 (3)	0.0534 (12)
C35	0.6600 (5)	0.0562 (4)	0.2639 (3)	0.0494 (10)
C36	0.6076 (5)	0.1445 (3)	0.3052 (2)	0.0403 (9)

Table 2. Selected geometric parameters (Å, °)

Au—C1	2.045 (4)	P—C31	1.817 (4)
Au—P	2.2859 (9)	C1—F1	1.318 (5)
P—C11	1.806 (4)	C1—F3	1.321 (5)
P—C21	1.810 (4)	C1—F2	1.333 (5)

C1—Au—P	178.56 (11)	C11—P—Au	113.92 (12)
C11—P—C21	106.6 (2)	C21—P—Au	113.38 (13)
C11—P—C31	104.9 (2)	C31—P—Au	110.45 (12)
C21—P—C31	107.0 (2)		

Program(s) used to solve structure: *SHELXTL-Plus* (Sheldrick, 1991). Program(s) used to refine structure: *SHELXL93* (Sheldrick, 1993).

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: AB1330). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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[Au(PPh₃)₂]⁺[BF₄]⁻

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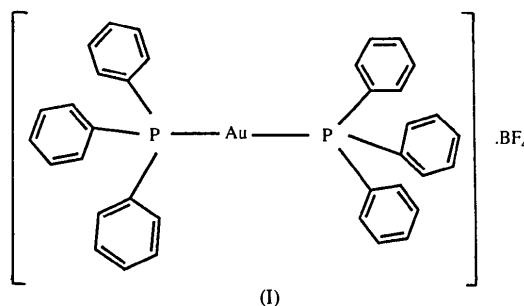
(Received 6 September 1993; accepted 4 September 1995)

Abstract

The synthesis and structural characterization of the title compound, bis(triphenylphosphine)gold(I) tetrafluoroborate, [Au(PPh₃)₂]⁺[BF₄]⁻, are reported. The BF₄⁻ anion does not interact with the Au centre. The coordination geometry about Au is close to linear, with Au—P distances of 2.321 (3) and 2.322 (3) Å, and a P1—Au—P2 angle of 167.3 (1)°. The range of P—C distances is 1.787 (10)–1.823 (9) Å.

Comment

A number of mononuclear (King, Khan, Staples & Fackler, 1992, and references therein) and binuclear (Wang, Garzon, King, Wang & Fackler, 1989; King, Wang, Khan & Fackler, 1989) gold(I) compounds are luminescent. Structural information is essential for understanding this luminescent property. In this paper we report the crystal structure of [Au(PPh₃)₂]⁺[BF₄]⁻, (I).



The coordination number of the central Au atom is two, with two P atoms linearly bound to it. All four F atoms in the BF₄⁻ anion are disordered. Two other [Au(PPh₃)₂]⁺X⁻-type structures have been reported recently (X = NO₃, PF₆; Staples, King, Khan, Winpenny & Fackler, 1993). Fig. 1 shows a drawing of the molecule with the atom-labelling scheme.

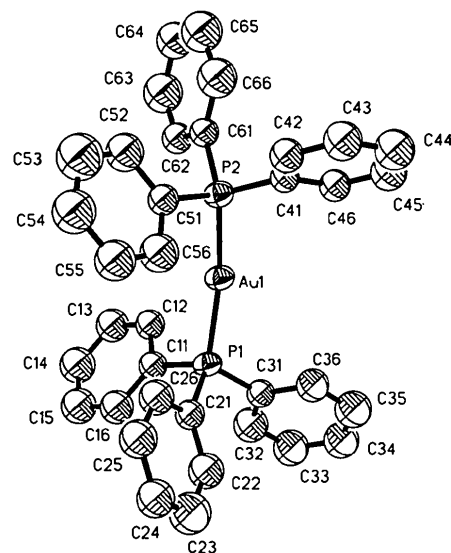


Fig. 1. Drawing (*SHELXTL-Plus*; Sheldrick, 1991) of the [Au(PPh₃)₂]⁺ cation showing 50% probability ellipsoids for non-H atoms.

Experimental

The title compound was synthesized by reacting [Au(PPh₃)₂]Cl with AgBF₄ in CH₂Cl₂ (DCM) and leaving overnight. After filtering off the precipitate, [Au(PPh₃)₂]BF₄ powders were formed by adding a large amount of diethyl ether (Et₂O) to the reaction solution. Colourless crystals of [Au(PPh₃)₂]BF₄ were grown from DCM/Et₂O mixed solvents.